

Structural Analysis of a Group of Phytoestrogens for the Presence of a 2-D Geometric Descriptor Associated with Non-genotoxic Carcinogens and Some Estrogens (44234)

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Abstract. Analysis of a group of phytoestrogens indicates that the majority of these chemicals are devoid of a 6Å distance descriptor biophore. This 6Å biophore is associated with the carcinogenicity of a set of chemicals in the mouse subset of the Carcinogenic Potency Database assembled by Gold *et al.* The prevalence of non-DNA-reactive carcinogens and chemicals endowed with estrogenic activity included in the group of chemicals possessing 6Å descriptor suggests that it describes a ligand binding site on an estrogen receptor. Evidence is presented that estrogens with and without the 6Å biophore bind to alternate receptors or to similar receptors but with different affinities. Since the 6Å biophore was identified based upon a carcinogenicity database, it is conceivable that binding to the putative receptor that recognizes this biophore is associated with carcinogenicity. Alternatively, estrogens devoid of this 6Å biophore may be noncarcinogenic, suggesting that carcinogenicity and estrogenicity may be separate phenomena.

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Phytoestrogens represent a diverse and naturally occurring source of estrogen exposure in animals and humans. Xenoestrogen exposure has been implicated in the etiology of breast cancer (1) as well as other hormonally related cancers. Alternatively, dietary intake of certain phytoestrogens has been suggested as diminishing the risk of certain cancers by their possible antiestrogenic action (see Ref. 2). The distinction between estrogen and antiestrogen may in fact be obscure. Several of the factors that determine the estrogenicity versus antiestrogenicity of a chemical are species- and tissue-specificity, and dose (reviewed in Ref. 3). Thus a debate exists over the possible risks and benefits from phytoestrogen exposure.

Traditional structure-activity relationship (SAR) analyses related to cancer causation have been based upon the somatic mutation and electrophilic theory of cancer causation (4). Detailed analyses of large cancer bioassay databases by Ashby and colleagues (5, 6) demonstrate the successful utilization of the electrophilic theory in identifying chemical substructures that are implicated in "genotoxic" carcinogenesis.

In fact, the majority of recognized human carcinogens are genotoxic (7, 8, 9). An exception to this generalization is found with hormonal carcinogens. Hormonal carcinogens are thought to act by a receptor-mediated mechanism (10, 11). However, some evidence also exists for the DNA-reactivity of diethylstilbestrol (12, 13) and tamoxifen (14, 15, 16, 17). This apparent multiplicity of action of certain hormonal carcinogens (i.e., estrogenic and genotoxic) must be considered when examining the possible risks (or benefits) of exposure to phytoestrogens.

The suitability of SAR and quantitative SAR (QSAR) involving non-genotoxic carcinogens is a matter of debate. The multiplicity of mechanisms believed to be involved in non-genotoxic carcinogenesis has been seen as making such an endeavor problematic (18, 19, 20, see also Ref. 21).

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However, in a recent study, we identified a 6Å geometric distance descriptor that appears to be associated with non-genotoxic carcinogens and may, in fact, be indicative of a ligand binding site for an estrogen receptor (22, 23). Additionally, estrogens with and without the 6Å biophore may bind to alternate receptors or with different affinities to the same receptors. Of course the possession of this biophore is only indicative of carcinogenicity and possible estrogenicity. Lipophilicity, steric constraints as well as other physico-chemical parameters will influence the binding ability and activities of these chemicals. The results contained herein represent excerpts and extensions of our work (22, 23) to phytoestrogens.

Methods

The computer automated structure evaluation system MULTICASE (MC) (24, 25) was used for these analyses. Basically, the system selects its own descriptors automatically from a learning set of chemicals composed of active and inactive molecules. The descriptors are continuous fragments that are embedded in the molecule. It is assumed that each fragment is not related to activity and will be randomly distributed between inactive and active molecules. Fragments that deviate from a random distribution are considered relevant to activity. The descriptors consist of either activating or inactivating fragments or geometric distance descriptors based on molecular lipophilic centers (see below). MC then selects the most important of these descriptors as a biophore (i.e., the functionality that is associated with the largest number of active molecules and fewest numbers of inactive molecules). These molecules then become a learning subset to identify the chemical and physico-chemical properties associated with their activity. This will result in a QSAR equation for this group of molecules. The

molecules explained here are removed from the set, and the process is iterated until all molecules are explained. Once the training set has been assimilated, MC can be queried regarding the predicted activity and potency of molecules of unknown activity.

MC incorporates the following rules to identify two-dimensional distance descriptors based upon the presence of lipophilic centers. These two-dimensional distances are calculated from the molecular structure. Heteroatoms and lipophilic carbon atoms are designated as "special" atoms. A carbon atom is designated as a lipophilic center if it is at least four bonds away from a heteroatom and is also the furthest carbon from the heteroatom when its neighbors are considered. After all the "special" atoms are selected, the distances between all possible pairings are calculated. The distribution of these descriptors among active and inactive molecules is analyzed for statistical significance. Various atom groupings are also investigated (i.e., hydrogen bond acceptors and donors as well as halogens).

The learning set used in this investigation was the Carcinogenic Potency Database (CPDB) assembled by Gold *et al.* (26, 27, 28, 29, 30). A mouse database was extracted from this compilation. The rationale for using a species-specific database as opposed to the entire CPDB is that a more coherent set of structural descriptors of carcinogenic activity may be expected in a more strictly defined system.

All dosages for chemicals reported were transformed into gavage equivalents (26). A carcinogenic potency value (TD₅₀) was calculated for each chemical. The calculated TD₅₀ is the dose required for 50% of the animals to remain cancer free throughout the course of the experiment (thus accounting for spontaneous cancer) (26, 31). Additionally, the reported TD₅₀ value in mg/kg/day was converted into mmol/kg/day. For the purpose of the SAR analyses, TD₅₀

Biophore A:

2D fragment [C-] <— 6.0 Å —> [OH-] conjugated and generic

14 out of the known 16 molecules (87%) containing such a biophore are mouse carcinogens with an average activity of 47. (conf.level= 100%)

Modulator 1: OH-CH-

Constant	51.8
Activating	34.2

The probability that this molecule is a mouse carcinogen is 83.3%

The compound is predicted to be extremely active
The projected mouse carcinogenicity activity 86 CASE units

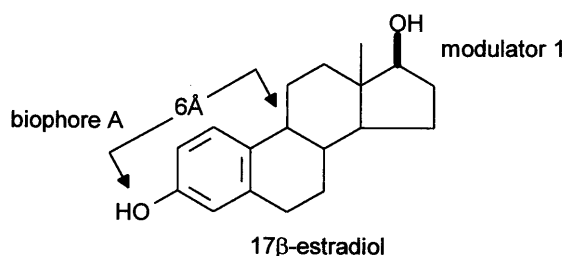


Figure 1. Prediction of the Murine Carcinogenicity of 17β-estradiol. An activity of 86 CASE units indicates a TD₅₀ value of 0.001 mmol/kg/day.

Table I. Distribution of the 6Å Distance Descriptor Biophore Among Estrogenic Chemicals

Chemical	Type	6Å
Chalcone	phytoestrogen	–
4-hydroxychalcone	phytoestrogen	–
4,4'-dihydroxychalcone	phytoestrogen	–
4-hydroxy-4'-methoxychalcone	phytoestrogen	–
2',4,4'-trihydroxychalcone (isoliquiritigenin)	phytoestrogen	–
2',4,4',6'-tetrahydroxychalcone (naringenin chalcone)	phytoestrogen	–
2',4,4',6'-tetrahydroxydihydrochalcone (phloretin)	phytoestrogen	–
flavone	phytoestrogen	–
3-hydroxyflavone	phytoestrogen	–
4'-hydroxyflavone	phytoestrogen	–
6-hydroxyflavone	phytoestrogen	–
7-hydroxyflavone	phytoestrogen	–
3',6-dihydroxyflavone	phytoestrogen	–
3',7-dihydroxyflavone	phytoestrogen	–
4',5-dihydroxyflavone	phytoestrogen	–
4',6-dihydroxyflavone	phytoestrogen	–
5,7-dihydroxyflavone (chrysin)	phytoestrogen	–
7,8-dihydroxyflavone	phytoestrogen	–
3',4',7-trihydroxyflavone	phytoestrogen	–
3,5,7-trihydroxyflavone (galangin)	phytoestrogen	–
4',5,7-trihydroxyflavone (apigenin)	phytoestrogen	–
4',7,8-trihydroxyflavone	phytoestrogen	–
3,3',4',7-tetrahydroxyflavone (fisetin)	phytoestrogen	–
3',4',5,7-tetrahydroxyflavone (luteolin)	phytoestrogen	–
3,4',5,7-tetrahydroxyflavone (kaempferol)	phytoestrogen	–
3,5,7-trihydroxy-4'-methoxyflavone (kaempferide)	phytoestrogen	–
3,3',4',5,7-pentahydroxyflavone (quercetin)	phytoestrogen	–
2',3,4',5,7-pentahydroxyflavone (morin)	phytoestrogen	–
3,3',4',5,5',7-hexahydroxyflavone (myricetin)	phytoestrogen	–
flavanone	phytoestrogen	–
4',7-dihydroxyflavanone	phytoestrogen	–
4',5,7-trihydroxyflavanone (naringenin)	phytoestrogen	–
3',5,7-trihydroxy-4'-methoxyflavanone (hesperetin)	phytoestrogen	–
3,3',4',5,7-pentahydroxyflavanone (taxifolin)	phytoestrogen	–
3,3',4',5,7-flavan pentol flavanone ([±] catechin)	phytoestrogen	–
isoflavone	phytoestrogen	–
4',7-dihydroxyisoflavone (diadzein)	phytoestrogen	–
7-hydroxy-4'-methoxyisoflavone (formononetin)	phytoestrogen	–
3',4',7-trihydroxyisoflavone	phytoestrogen	–
4',5,7-trihydroxyisoflavone (genistein)	phytoestrogen	–
4',6,7-trihydroxyisoflavone	phytoestrogen	–
5,7-dihydroxy-4'-methoxyisoflavone (biochanin A)	phytoestrogen	–
coumestrol	phytoestrogen	–
4,4'-dihydroxystilbene	phytoestrogen	+
3,5-dihydroxystilbene	phytoestrogen	–
coumarin	phytoestrogen	–
α-sitosterol	phytoestrogen	–
β-sitosterol	phytoestrogen	–
glucyrrhetic acid	phytoestrogen	–
zearalenol	phytoestrogen	–
zearalenone	phytoestrogen	–
α-zearalenol	phytoestrogen	–
indenestrol A	phytoestrogen	+
tetrahydrocannabinol	phytoestrogen	+
o,p'-DDE	xenoestrogen	–
chlordecone	xenoestrogen	–
diethylstilbestrol	estrogen	+
3'-hydroxy- <i>E</i> -diethylstilbestrol	estrogen	+
4',4'-diethylstilbestrol quinone	estrogen	–
tamoxifen	antiestrogen	–
3-hydroxytamoxifen	antiestrogen	–
4-hydroxytamoxifen acid	antiestrogen	+
toremifene	antiestrogen	–
4-hydroxy-deamino-hydroxytoremifene	antiestrogen	+
ICI 164,384	antiestrogen	+
ICI 182,780	antiestrogen	+
LY 117018	antiestrogen	–
MER 25	antiestrogen	–
17 β-estradiol	estrogen	+
17α-ethinyl estradiol	estrogen	+
benzestrol	estrogen	+
dienestrol	estrogen	+
estriol	estrogen	+
estrone	estrogen	+
hexestrol	estrogen	+
megestrol	estrogen	–
norgestrol	estrogen	–
norlestrin (isomer)	estrogen	–
phenol red	xenoestrogen	+

values were transformed into CASE activity units using Equation 1.

Chemicals were also assigned to activity groups. Chemicals reported by the authors to be non-carcinogenic in mice were assigned 10 CASE units together with chemicals exhibiting TD₅₀ values in excess of 51 mmol/kg/day. The chemicals in the range of 10–19 CASE units were designated as inactive. Chemicals with activities in the range of 20–29 CASE units were designated marginally active carcinogens, and chemicals in the range of 30–99 CASE units were considered carcinogenic. Overall, the mouse CPDB consisted of 636 chemicals, 291 of which were active, 11 marginal, and 334 noncarcinogenic.

$$\text{CASE activity} = 14.133 \times \log (1/\text{TD}_{50}) + 44.133 \quad (\text{Equation 1})$$

Results and Discussion

The 6Å biophore was identified as being associated with carcinogenicity in the mouse CPDB. Among the chemicals in the database that possessed the 6Å biophore were estradiol and related chemicals (Fig. 1), suggesting that this biophore may be related to the estrogenicity as well as the carcinogenicity of chemicals that contain it (21). In our original study, a series of 42 chemicals reported to be endowed with estrogenic activity were tested for the presence of the 6Å biophore; of these, about one-half contained the 6Å distance descriptor (see Table I for an abbreviated list).

The therapeutic antiestrogen tamoxifen lacked the 6Å biophore; however, its active antiestrogenic metabolite, 4-hydroxytamoxifen, possessed it. Oxidative metabolism is often required for chemicals to gain their estrogenicity as is the case for tamoxifen (3). Polychlorinated biphenyls and methoxychlor have also been shown to require oxidative metabolism to exhibit estrogenicity (32), and, in fact, hydroxylated metabolites of polychlorinated biphenyls display the 6Å descriptor. Moreover, the antiestrogens ICI 164,384 and ICI 182,780 contain the 6Å biophore whereas LY 117018, another antiestrogen, lacks it. It has been suggested that LY 117018 and tamoxifen (32, 34) and LY 117018, ICI 182,780, and 164,384 (35) have different bases for their antiestrogenicity. Thus possession or lack of the 6Å biophore may be indicative of a dichotomy that exists among estrogens.

The present exercise entailed analysis for the 6Å biophore in a group of 54 phytoestrogens (Table I). As mentioned above, oxidative metabolism may be needed to transform estrogens (i.e., proestrogens) into active congeners. Many of the chemicals in this set were hydroxylated and methoxylated congeners of flavonoids. Thus if oxidative metabolism was involved in the conversion of these chemicals to carcinogens or estrogens, this set of chemicals may contain active congeners. However, only indenestrol A and 4,4'-dihydroxystilbene, an analog of diethylstilbestrol, pos-

sessed the 6Å descriptor (Table I). No flavonoids or other phytoestrogens tested contained the 6Å biophore.

Conclusion

We postulated that the 6Å descriptor is associated not only with murine carcinogenicity, but also with an estrogen-receptor ligand. This suggests that the phenomena of estrogenicity and carcinogenicity may result from a similar mechanism as is possibly the case for the reported carcinogenicity of tamoxifen. However, a large population of chemicals endowed with estrogenic activity, particularly phytoestrogens, are devoid of this biophore, suggesting that a structurally based dichotomy may exist in the estrogenic response.

In fact Baker (36) suggests that the estrogenicity of phytoestrogens is derived from their inherent ability to interact with mammalian enzymes involved in the regulation and production of endogenous estrogens. Thus phytoestrogens may exert their estrogenicity through a mechanism that is not involved with carcinogenicity as indicated by the absence of the 6Å descriptor associated with carcinogenicity in mice.

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